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Synthesis and Use of 6-, 7-and 8-Membered P-Heterocycles

GYÖRGY KEGLEVICH^a, LÁSZLÓ TÖKE^a, KINGA STEINHAUSER^a, TIBOR NOVÁK^a and KRISZTINA LUDÁNYI^b

^aDepartment of Organic Chemical Technology, Technical University of Budapest, 1521 Budapest, Hungary and ^bInstitute of Chemistry of the Hungarian Academy of Sciences, 1025 Budapest, Hungary

The synthesis of new families of P-heterocycles by reductive modifications, ring enlargement and Diels-Alder cycloaddition is described.

Keywords: P-heterocycles; ring enlargement; reductive methods; fragmentations

In this paper, the latest results of ours in P-heterocyclic chemistry are summarized.

1. Synthesis of six-membered P-heterocycles by reductive modifications

It was observed that the catalytic hydrogenation of phosphabicyclohexanes 1 led to the diastereomers of hexahydrophosphinine oxides 2^[1]. The formation of product 2 is the result of a consecutive series of reactions involving intermediates 3-5. Decreasing the activity of the catalyst, we could achieve that monochloro species 3 should be the main component^[2].

Cl Cl Me
$$\frac{87 \, ^{\circ}\text{C}, \, 9 \, \text{bar}}{3 \, \text{Hz}, \, \text{Pd/C}}$$

$$2 \, \text{NB}_{3}, \, \text{ROH}$$

$$Y = \text{Ph, n-BuO, EtO, n-PrO, i-PrO}$$

$$\text{red.}$$

$$R^{1} \longrightarrow R^{2} \longrightarrow R^{1} \longrightarrow R^{2} \longrightarrow R^{2} \longrightarrow R^{1} \longrightarrow R^{2} \longrightarrow R^$$

We found that double-bonds with electron-withdrawing substituents at one end can be easily reduced by the borane-dimethyl sulfide (BMS) reagent. Thus, the reaction of the 2,3-dihydro-1H-phosphole oxides with BMS followed by hydrolysis afforded the corresponding tetrahydrophosphole oxides. Similar reaction of the 1,2-dihydrophosphinine oxides (6) gave the 1,2,3,6-tetrahydrophosphinines (7) in a regioselective manner^[3]. The reductions take place certainly through hydroboration. Both the sterie and the electronic reasons stand for the enhanced reactivity of the α,β -double bond.

$$R^{1} \xrightarrow{C1} R^{2} \xrightarrow{1.,24-40 \text{ °C}} \frac{2.3 \text{ eq. Me}_{2}\text{S} \cdot \text{BH}_{3}}{2.,\text{H}_{2}\text{O}} \xrightarrow{R^{1}} \frac{R^{2}}{\text{OP}_{Y}} \xrightarrow{A} \frac{R^{1}}{\text{A}} \frac{R^{2}}{\text{Me}} \xrightarrow{H} \frac{R^{2}}{\text{Me}}$$

$$Y = \text{Ph, MeO, EtO, i-PrO}$$

It was interesting to observe that the reaction of the phosphole-oxide dimers (8) with BMS resulted in a change in the functionality of the bridging P-center rather than selective reduction of the double-bond in the 2,3-dihydrophosphole moiety.

In the first step, the bridging phosphine oxide moiety in 8 is deoxygenated by the borane, then the phosphine so obtained is complexed by another unit of borane to give phosphine-borane 9.

Synthesis of seven-membered P-heterocycles by ring enlargement of tetrahydrophosphinine oxides

We wished to utilize the tetrahydrophosphinine oxides in the syntesis of seven-membered P-heterocycles. The reaction of tetrahydrophosphinine oxides 10 with dichlorocarbene afforded phosphabicycloheptanes 11. In the reaction of the 4-chloro derivative (10, X=CI) ca. 30% of 2,3-dihydrophosphepine oxide 12 was also formed. The thermolysis

of the dichlorocarbene adducts (11) led, in both cases, to the same 2,7-dihydro-phosphepine oxide (13)¹⁴.

Applying the dichlorocarbene ring enlargement to 3-methoxy-tetrahydrophosphinines 14, the reaction followed an unexpected route: 2,3,6,7-tetrahydrophosphepine oxides 15 could be isolated from the reaction mixture.

The key-step for the formation of products 15 is the isomerisation of 1,2,3,6-tetrahydrophosphinines 14 to 1,2,3,4-tetrahydro derivatives 16. The next step may involve Michael-type addition of a trichloromethyl anion at the end of the double-bond of 16 followed by cyclopropanation. Products 15 are formed by the opening of the cyclopropane ring in 17 by the effect of a chloride anion.

14
$$\xrightarrow{\tau_{H}\Theta}$$
 MeO $\xrightarrow{R^1}$ $\xrightarrow{R^2}$ $\xrightarrow{I,\Theta \subset G_3}$ MeO $\xrightarrow{R^1}$ $\xrightarrow{R^2 \subset I}$ $\xrightarrow{\pm_{G}\Theta}$ 15

 Synthesis of 2-phosphabicyclo[2.2.2]octene derivatives and their utilization in phosphorylations

We have synthesized Diels-Alder cycloadducts 18 and 19 to utilize them in the UV lightmediated phosphorylation of alcohols. The alcohols were added to the reaction mixture prior to irradiation.

$$R^{1}$$
 R^{1}
 R^{2}
 R^{1}
 R^{2}
 R^{1}
 R^{2}
 R^{1}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{4}
 R^{2}
 R^{4}
 R^{2}
 R^{4}
 R^{4

Phosphinic and phosphonic esters 20 were obtained in good yields. The mechanistic investigations suggest that beside the E-A pathway involving methylene phosphine oxide (Y-P(O)(CH₂)) as the intermediate, the A-E route is also realized. According to this, the alcohol is added to the phosphoryl group of the phosphabicyclooctene (18 or 19) to form an intermediate with a pentacoordinate phosphorus atom¹⁵¹.

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